

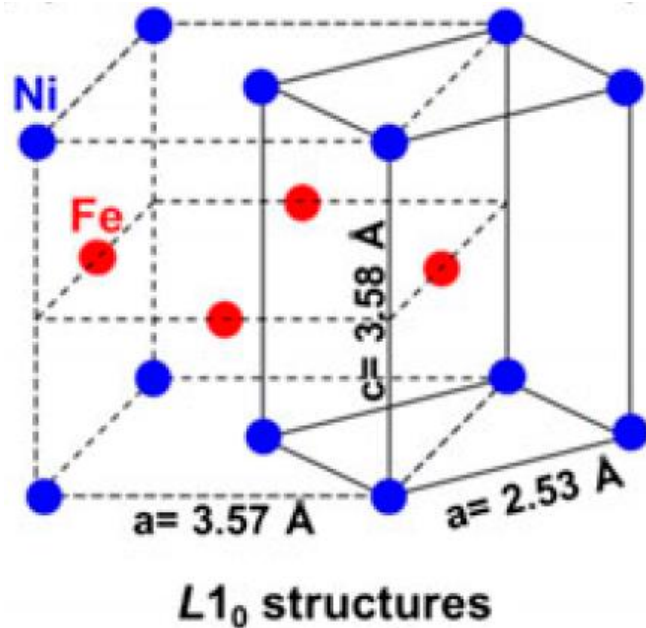
# $L_1O$ -FeNi alloy with addition composition

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## $L1_0$ structure FeNi



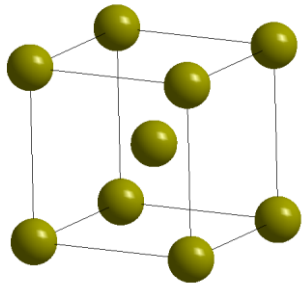
Tetragonal  $L1_0$ -ordered FeNi/tetrataenite/  $\gamma''$ , is a rare-earth-free magnetic compound.

saturation magnetization is  $1.6\mu\text{b}/\text{magnetic atom}$ .

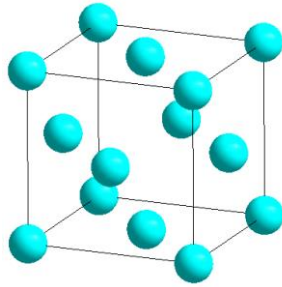
The compound naturally occurs in meteorites and requires millions of year to anneal in nature.

Theoretically magnetocrystalline anisotropy constant can be in the range  $K1=0.5\text{--}1\text{MJ}/\text{m}^3$ .

# Phases of Fe-Ni



Fe : bcc

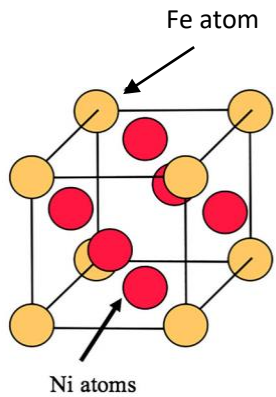


Ni: fcc

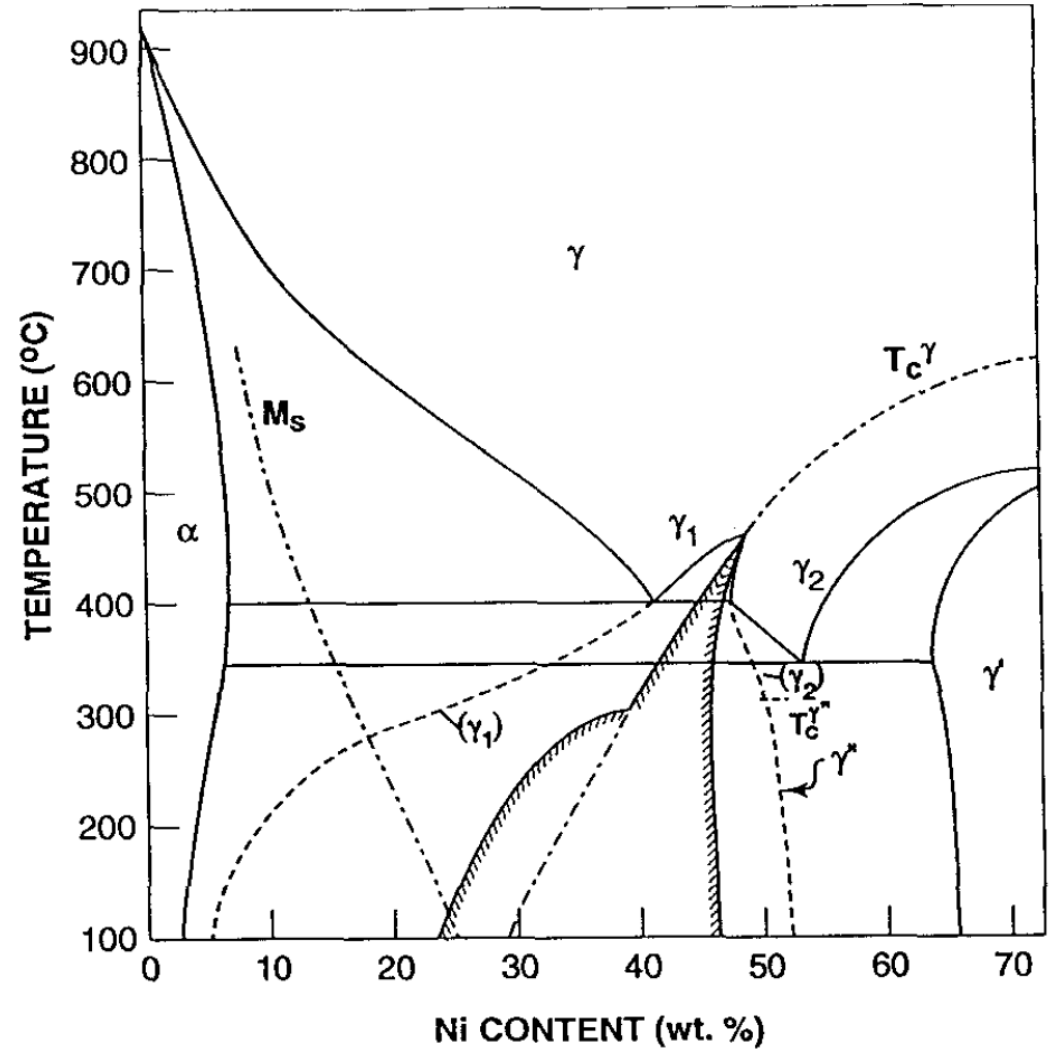
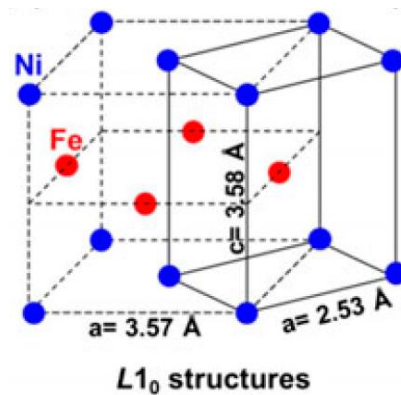
FeNi (any ratio)  $\alpha$  phase : well mixed bcc

FeNi (any ratio)  $\gamma$  phase : well mixed fcc

FeNi<sub>3</sub>  $\gamma'$  phase : L<sub>12</sub> structure



FeNi  $\gamma''$  phase : L<sub>10</sub> structure



Phase diagram of Fe-Ni

C.-W. Yang, D. B. Williams, and J. I. Goldstein, *J. Phase Equilib.* 17(6), 522 (1996).

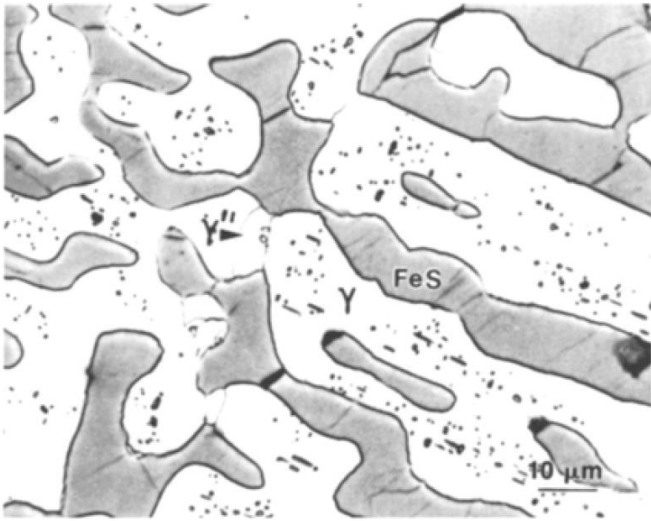
**Table 1 Fe-Ni-S Alloys and Heat Treatments**

| Temperature,<br>°C | Alloy      | Aging time,<br>days | Structure   |
|--------------------|------------|---------------------|---|
| 900.....           | FeNi2.5S10 | 15                  | $\gamma + \text{FeS}$                               |
|                    | FeNi5S10   | 15                  | $\gamma + \text{FeS}$                               |
|                    | FeNi10S10  | 15                  | $\gamma + \text{FeS}$                               |
|                    | FeNi20S10  | 10                  | $\gamma + \text{FeS}$                               |
| 800.....           | FeNi2.5S10 | 42                  | $\alpha + \gamma + \text{FeS}$                      |
|                    | FeNi5S10   | 42                  | $\gamma + \text{FeS}$                               |
|                    | FeNi10S10  | 42                  | $\gamma + \text{FeS}$                               |
|                    | FeNi20S10  | 42                  | $\gamma + \text{FeS}$                               |
|                    | FeNi30S10  | 42                  | $\gamma + \text{FeS}$                               |
| 700.....           | FeNi2.5S10 | 90                  | $\alpha + \text{FeS}$                               |
|                    | FeNi5S10   | 90                  | $\alpha + \gamma + \text{FeS}$                      |
|                    | FeNi10S10  | 90                  | $\gamma + \text{FeS}$                               |
|                    | FeNi20S10  | 90                  | $\gamma + \text{FeS}$                               |
|                    | FeNi30S10  | 110                 | $\gamma + \text{FeS}$                               |
| 600.....           | FeNi2.5S10 | 150                 | $\alpha + \text{FeS}$                               |
|                    | FeNi5S10   | 150                 | $\alpha + \gamma + \text{FeS}$                      |
|                    | FeNi10S10  | 150                 | $\alpha + \gamma + \text{FeS}$                      |
|                    | FeNi30S10  | 150                 | $\gamma + \gamma'' + \text{FeS}$                    |
| 500.....           | FeNi2.5S10 | 150                 | $\alpha + \text{FeS}$                               |
|                    | FeNi5S10   | 100                 | $\alpha + \gamma + \text{FeS}$                      |
|                    | FeNi10S10  | 100                 | $\alpha + \gamma + \text{FeS}$                      |
|                    | FeNi20S10  | 100                 | $\alpha + \gamma + \text{FeS}$                      |
|                    | FeNi30S10  | 206                 | $\gamma + \gamma'' + \text{FeS}$                    |
| 400.....           | FeNi2.5S10 | 165                 | $\alpha + \text{FeS}$                               |
|                    | FeNi5S10   | 165                 | $\alpha + \text{FeS}$                               |
|                    | FeNi10S10  | 165                 | $\alpha + \gamma + \text{FeS}$                      |
|                    | FeNi20S10  | 165                 | $\alpha + \gamma + \text{FeS}$                      |
|                    | FeNi30S10  | 165                 | $\gamma + \gamma' + \text{FeS}$                     |
| 300.....           | FeNi2.5S10 | 165                 | $\alpha + \text{FeS}$                               |
|                    | FeNi5S10   | 165                 | $\alpha + \text{FeS}$                               |
|                    | FeNi10S10  | 165                 | $\alpha + \text{FeS} + \alpha_2(a)$                 |
|                    | FeNi20S10  | 165                 | $\alpha + \gamma \text{ or } \gamma'' + \text{FeS}$ |
|                    | FeNi30S10  | 165                 | $\alpha + (\text{Fe,Ni})_9\text{S}_8 + \text{FeS}$  |

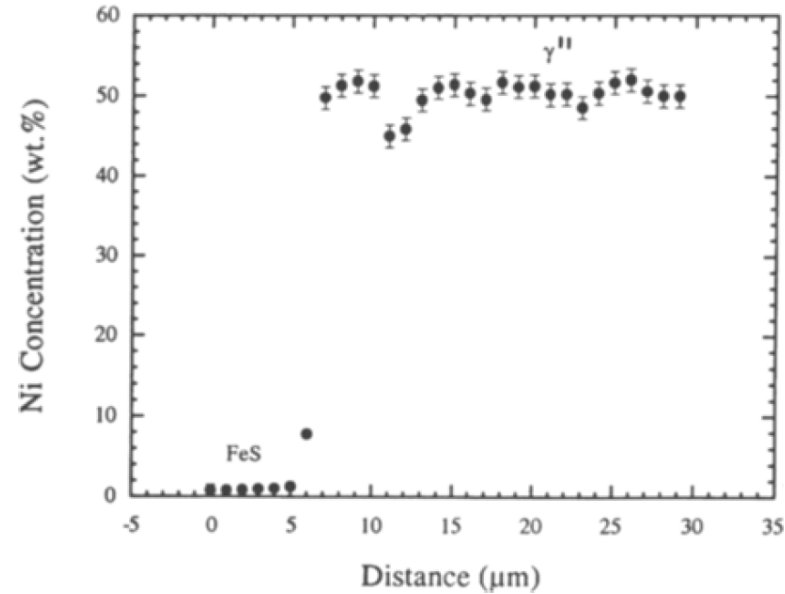
sulfur has been found to positively affect the L10 phase formation

(2.5, 5, 10, 20, and 30 wt.% Ni, 10 wt. % S, balance Fe)

Method: induction melting in alumina crucibles (Ar environment)



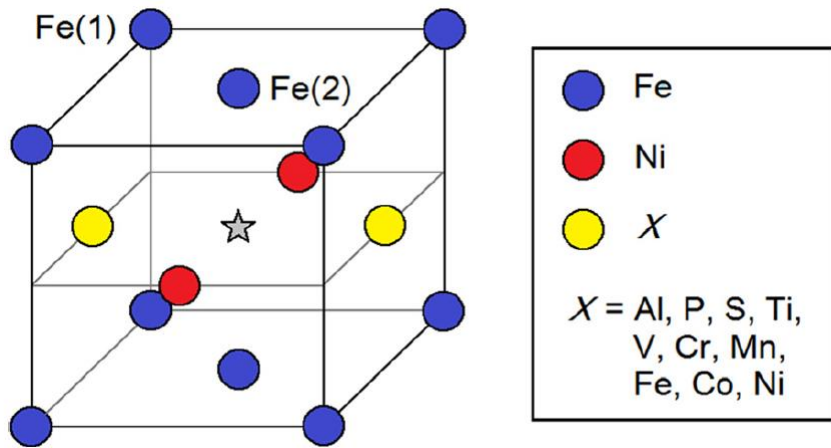
Light optical microscope image for FeNi30S10 in 500C



Ni composition profile across an FeS/  $\gamma''$  interface by electron probe microanalysis.

51.5±1.6 wt% in  $\gamma''$

Transmission electron microscopy SAD : $\gamma''$  has fcc structure with a lattice parameter of 0.35 nm.



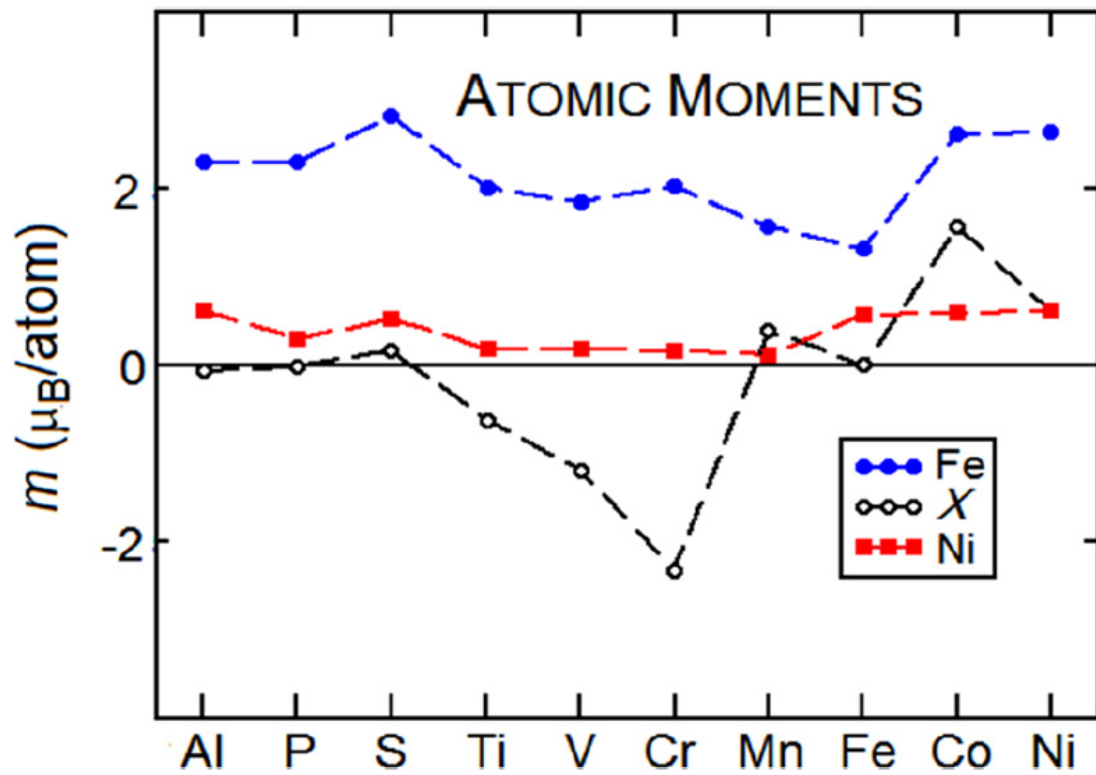
Basic  $L1_0$   $Fe_2NiX$  structures. The star in the center of the cell is the octahedral interstitial site in the Ni plane (no atom occupied).

TABLE I. Optimized lattice constants of the considered  $Fe_2XNi$  compounds.

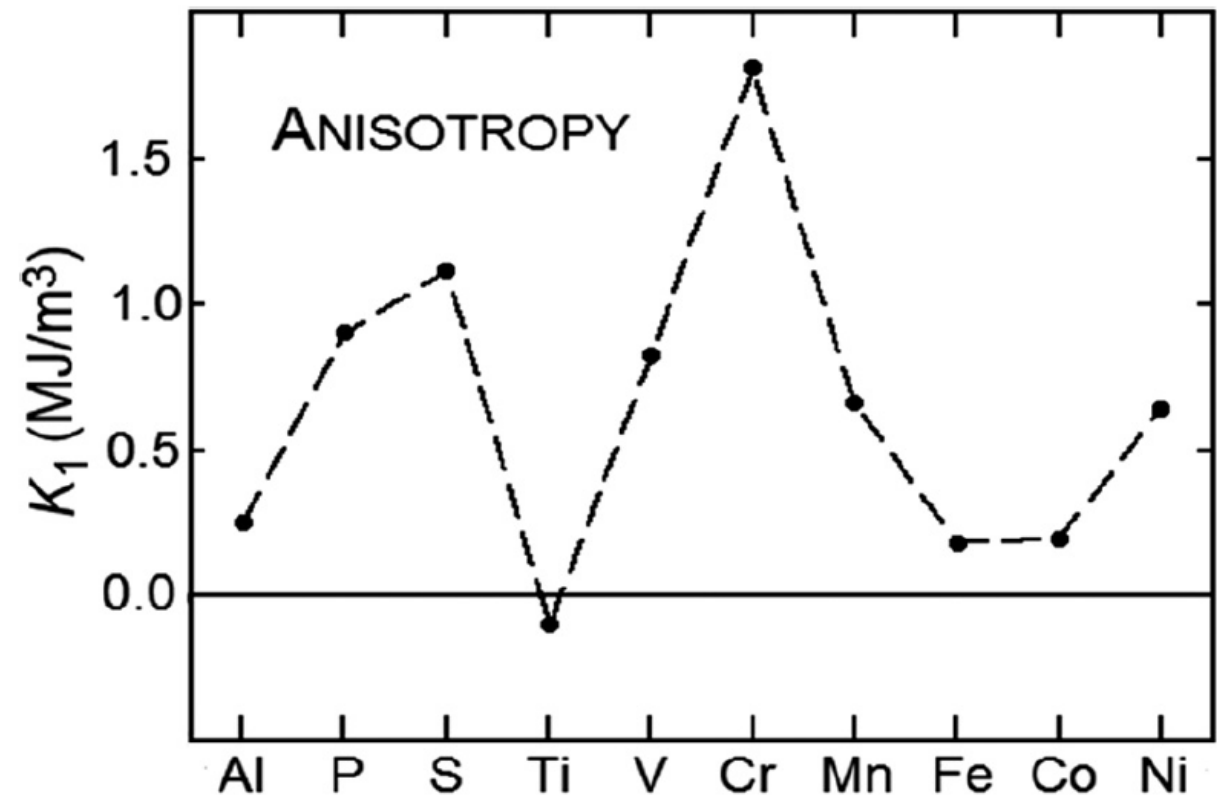
| X  | $a$ (Å) | $c$ (Å) |
|----|---------|---------|
| Al | 3.494   | 3.864   |
| P  | 3.457   | 3.754   |
| S  | 3.537   | 3.834   |
| Ti | 3.571   | 3.797   |
| V  | 3.577   | 3.559   |
| Cr | 3.561   | 3.571   |
| Mn | 3.504   | 3.506   |
| Fe | 3.594   | 3.594   |
| Co | 3.623   | 3.465   |
| Ni | 3.555   | 3.582   |

Optimized lattice constants of the considered  $Fe_2NiX$  compounds.

*Manchanda, P., Skomski, R., Bordeaux, N., Lewis, L., & Kashyap, A. Journal of Applied Physics* **115**, 17A710 (2014)



Atomic Moments in  $\text{Fe}_2\text{NiX}$



Magnetocrystalline anisotropies in  $\text{Fe}_2\text{NiX}$

Manchanda, P., Skomski, R., Bordeaux, N., Lewis, L., & Kashyap, A. *Journal of Applied Physics* **115**, 17A710 (2014)

# Conclusion

sulfur has been found to positively affect the L10 phase formation

For  $\text{Fe}_2\text{NiX}$  alloy,  $x=\text{S}$ , Cr may be an option of permanent magnet

The magnetism of the modified alloys will require further experimental and theoretical work