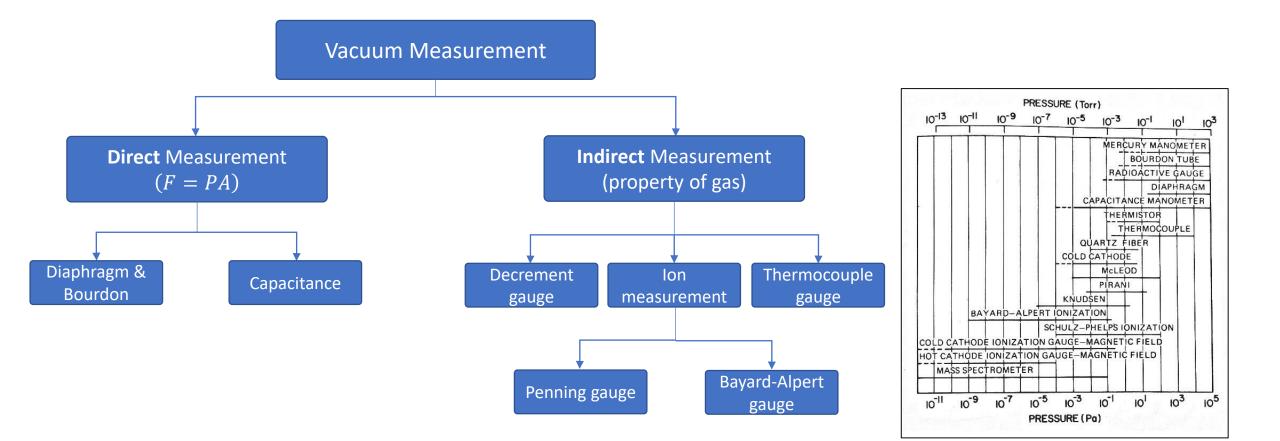
Vacuum Technology III: Pressure Measurement

Corbyn Mellinger

Xu Group Meeting

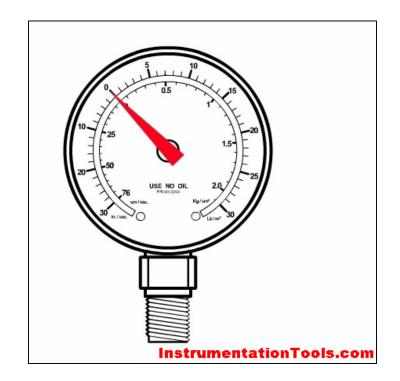
2018-05-25

Measurement Classifications



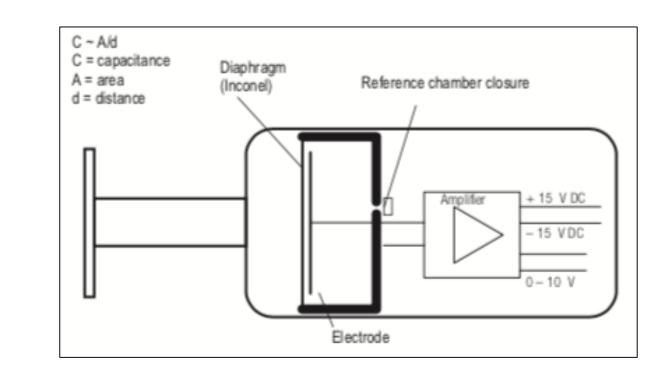
Diaphragm & Bourdon Gauge

- Pressure of a gas exerts a force, causing displacement
 - Measure this displacement to read out pressure
- Diaphragm: flexing of surface moves an indicator in a dial
- Bourdon: curved tube tries to extend when filled with gas, moves indicator on dial



Capacitance Gauge

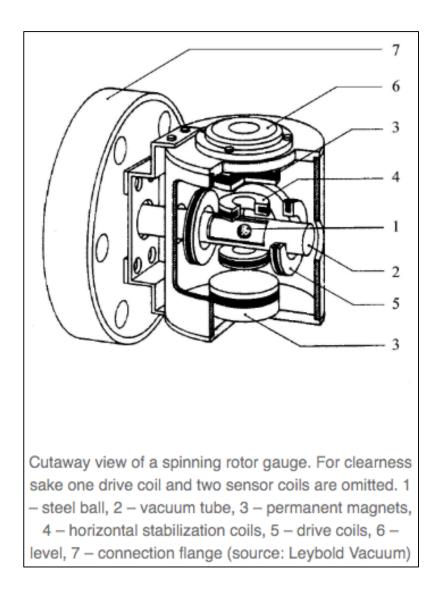
- Similar to diaphragm gauge, but flexible metal plate moves & changes capacitance
- Have to calibrate reference chamber using another gauge
- Some (but very little) dependence on gas type



Decrement Gauge

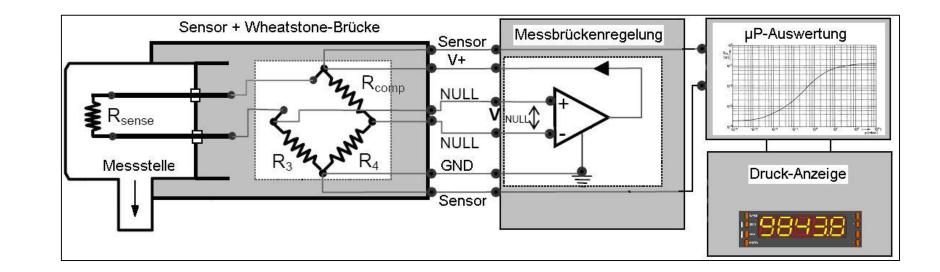
- Measure friction on a levitated steel ball
 - Speed reduction dependent on ball geometry, pressure of gasses

• $-f \frac{df}{dt} = \frac{10}{\pi} \frac{p \sigma}{\bar{c} r \rho}$; f is decline of speed; p is pressure; r, ρ , σ are related to ball geometry; \bar{c} is mean speed of gas (needs calibration)



Thermocouple Gauge

- Detects change in thermal conductivity of gas
 - "Pirani gauge" most common
- Heat loss of R_{sense} depends on pressure. Feedback to keep temperature constant \rightarrow measure of chamber pressure



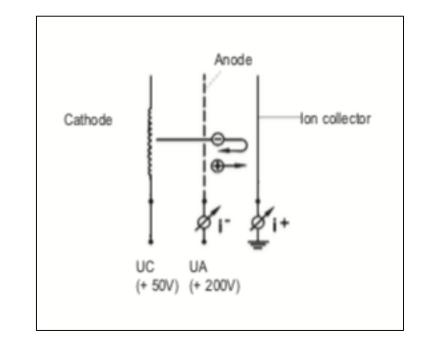
Cold Cathode Ion Gauge (Penning)

- Same configuration as ion pumping device from last week
 - Gasses ionized by free electrons, hit plate to register current
- Measure below ~10mTorr
- Pumping action results in relatively low accuracy

• Widely used due to its low cost and simple operation principles

Hot Cathode Ion Gauge (Bayard-Alpert)

- A hot cathode ionizes gas, ions move to anode to register current
- Two limiting effects:
 - X-ray effect: Hot electrons → inv. photoelectric effect → photoelectrons show as neg. current
 - Ion desorption effect: Electron impacts release adsorbed ions which flow to ion collector



Thanks